



Sulfuric Acid

1. PRODUCT AND COMPANY IDENTIFICATION

Product Name: Sulfuric Acid

Synonyms: Sulphuric Acid, Hydrogen Sulphate, Oil of Vitriol, Battery Acid

Product Use: Used in manufacture of fertilizers, explosives, other acids, metal pickling and petroleum processing.

Chemtrade Logistics Inc.

111 Gordon Baker Road
Suite 301
North York, ON
M2H 3R1
(866) 887-8805
www.chemtradelogistics.com

Emergency Telephone Number

Chemtrec 1-800-424-9300

Canutec (613) 996-6666

2. COMPOSITION/INFORMATION ON INGREDIENTS

<u>Hazardous Ingredients</u>	<u>% by Wt.</u>	<u>CAS Number</u>
Sulfuric Acid	70-100%	7664-93-9
<u>Non-Hazardous Ingredients</u>		
Water	0-30%	7732-18-5

3. HAZARD INFORMATION

EMERGENCY OVERVIEW:

Danger! Extremely corrosive. Causes severe burns and eye damage. Mist: Causes respiratory irritation. Harmful if inhaled. Harmful or fatal if swallowed. Reacts violently with water. Concentrated Sulfuric Acid will react with many organic materials and may cause fire due to the heat of the reaction. Not flammable, but reacts with most metals to form explosive/flammable hydrogen gas. Read the entire MSDS for a more thorough evaluation of the hazards.

National Fire Protection Association (NFPA) Rating

Hazardous Materials Identification System (HMIS) Rating

	NFPA	HMIS	4 = Extreme/Severe
Health	3	3	3 = High/Serious
Fire	0	0	2 = Moderate
Reactivity	2	2	1 = Slight
Special	W		0 = Minimum
Personal		C	W = Water Reactive
Protection			

Sulfuric Acid

3. HAZARD INFORMATION (continued)

Exposure Limits:

	ACGIH (TLV)	OSHA (PEL)	NIOSH (REL)
Sulfuric Acid	1 mg/m ³ (TWA) 3 mg/m ³ (STEL) A2 (Notations) refers to sulfuric acid contained in strong inorganic acid mists; suspected human carcinogen	1 mg/m ³ (TWA)	1 mg/m ³ (TWA)

POTENTIAL HEALTH EFFECTS:

Eye Contact: Immediate pain, severe burns and corneal damage, which may result in permanent blindness.

Skin Contact: Causes burns, and brownish or yellow stains. Concentrated solutions may cause second or third degree burns with severe necrosis. Prolonged and repeated exposure to dilute solutions may cause irritation, redness, pain and drying and cracking of the skin.

Inhalation: Causes respiratory irritation and at high concentrations may cause severe injury, burns, or death. Effects of exposure may be delayed.

Ingestion: Causes severe irritation or burns of the mouth, throat, and esophagus.

Existing Medical Conditions Possibly Aggravated By Exposure: Skin irritation may be aggravated in individuals with existing skin lesions. Breathing of vapors or sprays (mists) may aggravate acute or chronic asthma and chronic pulmonary disease such as emphysema and bronchitis.

Carcinogenicity: Strong inorganic acid mists containing sulfuric acid (Occupational exposures) : Proven (Human, Group 1, IARC) ; Suspected (Human Group A2, ACGIH) ; Group 3 (NTP) ; Classification not applicable to sulfuric acid and sulfuric acid solutions.

Sulfuric Acid

4. FIRST AID MEASURES

General: Corrosive effects on the skin and eyes may be delayed, and damage may occur without the sensation or onset of pain. Strict adherence to first aid measures following any exposure is essential. SPEED IS ESSENTIAL. OBTAIN IMMEDIATE MEDICAL ATTENTION.

Skin Contact: Immediately flush skin with running water for a **minimum** of 20 minutes. Start flushing while removing contaminated clothing. If irritation persists, repeat flushing. Obtain medical attention IMMEDIATELY. Do not transport victim unless the recommended flushing period is completed or flushing can be continued during transport.

While the patient is being transported to a medical facility, apply compresses of iced water. If medical treatment must be delayed, immerse the affected area in iced water. If immersion is not practical, compresses of iced water can be applied. Avoid freezing tissues.

Discard heavily contaminated clothing and shoes in a manner that limits further exposure. Otherwise, wash clothing separately before reuse.

Eye Contact: Immediately flush eyes with running water for a **minimum** of 20 minutes. Hold eyelids open during flushing. If irritation persists, repeat flushing. Obtain medical attention IMMEDIATELY. Do not transport victim until the recommended flushing period is completed unless flushing can be continued during transport.

Inhalation: Move victim to fresh air. Give artificial respiration ONLY if breathing has stopped. Do not use mouth-to-mouth method if victim ingested or inhaled the substance: induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Give Cardiopulmonary Resuscitation (CPR) if there is no pulse AND no breathing. Obtain medical attention IMMEDIATELY.

Ingestion: DO NOT INDUCE VOMITING. If victim is alert and not convulsing, rinse mouth and give ½ to 1 glass of water to dilute material. If spontaneous vomiting occurs, have victim lean forward with head down to avoid breathing in of vomitus, rinse mouth and administer more water. IMMEDIATELY contact local poison control center. Vomiting may need to be induced but should be directed by a physician or a poison control centre. IMMEDIATELY transport victim to an emergency facility.

Note to Physicians: This product contains materials that may cause severe pneumonitis if aspirated. If ingestion has occurred less than 2 hours earlier, carry out careful gastric lavage; use endotracheal cuff if available, to prevent aspiration. Observe patient for respiratory difficulty from aspiration pneumonitis. Give artificial resuscitation and appropriate chemotherapy if respiration is depressed. Following exposure the patient should be kept under medical review for at least 48 hours as delayed pneumonitis may occur. DO NOT attempt to neutralize the acid with weak bases since the reaction will produce heat that may extend the corrosive injury

Sulfuric Acid

5. FIRE AND EXPLOSION DATA

Flash Point (method): Not applicable. Not combustible

Flammable Limits (Lower): Not applicable.

Flammable Limits (Upper): Not applicable.

Auto Ignition Temperature: Not applicable.

Combustion and Thermal Decomposition Products: Oxides of Sulfur

Fire and Explosion Hazards: Not flammable but highly reactive. Strong dehydrating agent, which may cause ignition of finely divided combustible materials on contact. Reacts violently with water with evolution of heat can react with organic materials explosively (See Section 10). Reacts with many metals to liberate hydrogen gas which can form explosive mixtures with air. Hydrogen, a highly flammable gas, can accumulate to explosive concentrations inside drums, or any types of steel containers or tanks upon storage. Oxides of sulfur may be produced in fire.

Fire Fighting Instructions: Wear a NIOSH/MSHA approved self-contained breathing apparatus if vapors or mists are present and full protective clothing. For fighting fires in close proximity to spill or vapors, use acid-resistant personal protective equipment. Evacuate personnel to a safe area. Prevent unauthorized entry to fire area. Dike area to contain runoff and prevent contamination of water sources. Neutralize runoff with lime, soda ash or other suitable neutralizing agents (see Deactivating Chemicals, Section 6). Cool containers that are exposed to flame with streams of water until fire is out.

NOTE: Also see "Section 10 - Stability and Reactivity"

6. ACCIDENTAL RELEASE MEASURES

Steps to be taken in the event of a spill or leak: Remove all ignition sources (no smoking, flares, sparks or flames). All equipment should be grounded. Ventilate area. Use appropriate Personal Protection Equipment. Prevent liquid from entering sewers or waterways. Stop or reduce leak if safe to do so.

Small Spills: Cover with DRY earth, sand or other non-combustible material. Use clean non-sparking tools to collect material and place it into loosely covered plastic containers for later disposal.

Large Spills: Prevent liquid from entering sewers or waterways. Dike with inert material (sand, earth, etc.). Collect into plastic containers for disposal. Consider insitu neutralization and disposal. Ensure adequate decontamination of tools and equipment following clean up. Comply with Federal, Provincial/State and local regulations on reporting releases.

Waste Disposal Methods: Dispose of waste material at an approved waste treatment/disposal facility, in accordance with applicable regulations. Do not dispose of waste with normal garbage or to sewer systems.

Note - Clean-up material may be a RCRA Hazardous Waste on disposal.

- Spills are subject to CERCLA reporting requirements: RQ = 1000 lbs.

Sulfuric Acid

7. HANDLING AND STORAGE

Handling: Wear appropriate Personal Protection Equipment. Do not breathe sprays or mists. Do not ingest. Do not get in eyes, on skin or on clothing. Keep ignition sources away from sulfuric acid storage, handling and transportation equipment. Locate safety shower and eyewash station close to chemical handling area. Use EXTREME care when diluting with water. **Always add acid to water.** CAUTION: Hydrogen, a highly flammable gas, can accumulate to explosive concentrations inside drums, or any types of steel containers or tanks upon storage. **Carbon steel storage tanks must be vented.** People working with this chemical should be properly trained regarding its hazards and its safe use.

Storage: If stored in non-reactive container, keep container tightly closed. Metal and, specifically carbon steel, storage tanks must be vented due to hydrogen release as noted above.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Engineering Controls: Local exhaust ventilation should be applied wherever there is an incidence of point source emissions or dispersion of regulated contaminants in the work area. Ventilation control of the contaminant as close to its point of generation is both the most economical and safest method to minimize personnel exposure to airborne contaminants. The most effective measures are the total enclosure of processes and the mechanization of handling procedures to prevent all personal contact with sulfuric acid. Electrical installations should be protected against the corrosive action of acid vapors. Smoking should be prohibited in areas in which sulfuric acid is stored or handled.

Respiratory Protection: A NIOSH/MSHA approved air-purifying respirator equipped with acid gas/fume, dust, mist cartridges for concentrations up to 10 mg/m³. An air-supplied respirator if concentrations are higher or unknown.

Skin Protection: RECOMMENDED: Impervious (i.e., neoprene, PVC) gloves, coveralls, boots and/or other acid resistant protective clothing.

Eye Protection: Tight-fitting chemical goggles and face shield.

Other Personal Protective Equipment: Where there is a danger of spilling or splashing, acid resistant aprons or suits should be worn. Trouser legs should be worn outside (not tucked in) rubber boots. Safety showers and eyewash fountains should be installed in storage and handling areas.

Pictograms



EXPOSURE GUIDELINES:

HAZARDOUS INGREDIENT(S):

Sulfuric Acid:

ACGIH TLV
ACGIH STEL
OSHA PEL
NIOSH REL

1 mg/m³ (TWA) 8 hours
3 mg/m³ 15 minutes
1 mg/m³ (TWA) 8 hours
1 mg/m³ (TWA) 10 hours



9. PHYSICAL AND CHEMICAL PROPERTIES

Molecular Weight: 98.08

Physical State: Liquid

Appearance and Odor: Odourless, clear to amber, heavy, oily liquid. A pungent odor may exist if certain impurities are present in the acid.

Odor Threshold: Not applicable

Boiling Point: 77.67%: 193°C (380°F); 93.19%: 276°C (529°F); 98%: 330°C (626°F)

Melting/Freezing Point: 77.67%: -11.2°C (+11.6°F); 93.19%: -29.5°C (-21.1°F); 98%: -1.1°C (30°F)

Vapor Pressure at 40°C (102°F): 77.67%: 1.2 mmHg; 93.19%: 0.0016 mmHg; 98%: 0.002 mmHg

Specific Gravity at 15°C (60°F): 77.67%: 1.7059; 93.19%: 1.8354; 98%: 1.8437

Vapor Density: (Air=1): 3.4 sulfuric acid component

Bulk Density: Not applicable (see specific gravity)

Evaporation Rate: Not applicable

Solubility: Miscible in all proportions in water.

pH: 0.3 (1N solution at 25°C/78°F)

10. STABILITY AND REACTIVITY

Stability: The product is stable under normal conditions.

Conditions to Avoid: Keep away from heat and sources of ignition. Avoid temperatures, which may have a negative effect on the materials of construction used in equipment.

Materials to Avoid: Contact with organic materials (such as alcohol, acrylonitrile, chlorates, carbides, epichlorohydrin, fulminates, isoprene, nitrates and picrates) may cause fire and explosions. Contact with metals may produce flammable hydrogen gas. When diluting, add acid to water. Do NOT add water to the acid.

Hazardous Decomposition or Combustion Products: Toxic gases and vapors (e.g. sulfur dioxide, sulfuric acid vapors/mists and sulfur trioxide) may be released when sulfuric acid decomposes.

Hazardous Polymerization: Will not occur

Corrosivity: The product is corrosive.



11. TOXICOLOGICAL INFORMATION

Toxicological Data: LD₅₀ (oral, rat) = 2140 mg/kg
LC₅₀ (inhalation, rat) = 510 mg/m³ for 2 hrs
Skin effects (rabbit): Severe irritation
Eye effects (rabbit): Severe irritation

Carcinogenicity Data: The International Agency for Research on Cancer (IARC) has concluded that occupational exposure to strong inorganic acid mists containing sulfuric acid is carcinogenic to man, causing cancer of the larynx (the voice box). Although no direct link has been established between exposure to sulfuric acid, itself, and cancer in man, exposure to any mist or aerosol during the use of this product should be avoided See Section 3. Hazard Information, regarding Potential Health Effects (Long Term Exposure) for further discussion.

The National Toxicology Program (NTP) does not classify sulfuric acid or strong inorganic acid mists as known (or reasonably anticipated to be) human carcinogens.

Reproductive Effects: Slightly embryotoxic in rabbits (a minor, rare skeletal variation). The animals were exposed to 5 and 20 mg/m³ for 7 hrs/day throughout pregnancy. Slight maternal toxicity was present at the highest dose in both species.

Mutagenicity Data: Cytogenic analysis (hamster) ovaries 4 mmol/L

Teratogenicity Data: Not teratogenic in mice and rabbits.

Synergistic Materials: None known

12. ECOLOGICAL INFORMATION

Ecotoxic Effects: Harmful to aquatic life in very low concentrations. May be dangerous if it enters water intake; Fish toxicity; 2.8 µg/L 96 hrs LC50 Rainbow trout.

Products of Degradation: These products are sulphur oxides (SO₂, SO₃)

Toxicity of the Products of Degradation: The products of degradation are more toxic than the original product.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: Cleaned up material may be a hazardous waste as defined by Resource Conservation and Recovery Act (RCRA) on disposal due to the corrosivity characteristic. DO NOT flush to surface water or sanitary sewer system. Waste must be disposed of in accordance with federal, state, provincial and local environmental control regulations.



Sulfuric Acid

14. TRANSPORT INFORMATION

U.S. (Under DOT)

Shipping Name: Sulfuric acid
Hazard Class or Division: 8
Identification No.: UN1830
Packing Group: II
Reportable Quantity = 1000 pounds (454 kg)
IMO: 8
IATA/ICAO Class: 8

ER Guide: 137

Canada (Under TDG)

Shipping Name: Sulfuric acid
Classification(s): Class 8
Product Identification No. (PIN): UN1830
Packing Group: II
Regulated Limit: 50 kg

15. REGULATORY INFORMATION

U.S.A.

SARA Title III HAZARD CATEGORIES AND LISTS

Product Hazard Categories

Acute (Immediate) Health: Yes
Chronic (Delayed) Health: Yes
Fire: No
Reactivity: Yes
Sudden Release of Pressure: No

Lists

Extremely Hazardous Substance Yes
(40 CFR 355, SARA Title III Section 302)
CERCLA Hazardous Substance Yes
(40 CFR 302.4)
Toxic Chemical Yes
(40 CFR 372.65, SARA Title III Section 313)

Reportable Quantity (RQ) under U.S. EPA CERCLA: RQ=1000 lb / 454 kg

TSCA Inventory Status: Reported/Included

Right-To-Know: Illinois, Massachusetts, New Jersey, Pennsylvania

Other Regulations/Legislation which apply to this product: New Jersey Special Health Hazard Substance List and Environmental Hazardous Substance; Minnesota, Florida, Rhode Island Hazardous Substance ; California Director's List of Hazardous Substances; Massachusetts Extraordinarily Hazardous Substance List



15. REGULATORY INFORMATION (continued)

CANADA

Workplace Hazardous Materials Information System (WHMIS)

WHMIS Classification(s): Class D1A - Very Toxic
Class D2B – Suspected Human Carcinogen
Class E - Corrosive

WHMIS Health Effects Index: Acute Lethality - very toxic – immediate
Materials Causing Other Toxic Effects- Chronic
Corrosive to animal skin

WHMIS Ingredient Disclosure List: Confirmed A; Meets criteria for disclosure at 1% or greater.

National Pollutant Release Inventory (NPRI): Included

European:

EEC CLASSIFICATION: C, R 35

EINECS: 231-639-5

16. OTHER INFORMATION

REFERENCES:

1. **RTECS-Registry of Toxic Effects of Chemical Substances**, On-line search, Canadian Centre for Occupational Health and Safety RTECS database, Doris V. Sweet, Ed., National Institute for Occupational Safety and Health, U.S. Dept. of Health and Human Services, Cincinnati, Entry Update/Dec1997.
2. ChemAdvisor, Canadian Centre for Occupational Health and Safety, October 1998.
3. **CHEMINFO**, through "**CCINFODisc**", Canadian Centre for Occupational Health and Safety, Hamilton, Ontario, Canada, (August, 1998).
4. **Hazardous Substances Data Bank** , through "CCINFO disc", Canadian Centre for Occupational Health and Safety, Hamilton, Ontario, Canada, (2007).
5. NIOSH POCKET GUIDE TO CHEMICAL HAZARDS, U.S. Department of Health and Human Services, National Institute for Occupational Safety and Health, 2006.
6. "8th Report on Carcinogens – 1998 Summary", National Toxicology Program.
7. Sax, N.I., "Dangerous Properties of Industrial Materials", 7th Edition, 1989.
8. "2006 Threshold Limit Values and Biological Exposure Indices", American Conference of Government Industrial Hygienists, 2006.
9. Merck, 12th Edition, 1999.
10. Supplier's Material Safety Data Sheets.



16. OTHER INFORMATION (continued)

Legend:

AFFF	- Aqueous Film Forming Foam
CAS #	- Chemical Abstracts Service Registry Number
CERCLA	- Comprehensive Environmental Response, Compensation, and Liability Act
CFR	- Code of Federal Regulations
DOT	- Department of Transportation
EPA	- Environmental Protection Agency
LC ₅₀	- The concentration of material in air expected to kill 50% of a group of test animals
LD ₅₀	- Lethal Dose expected to kill 50% of a group of test animals
LEL	- Lower Explosive Limit
MSHA	- Mine Safety and Health Administration
NIOSH	- National Institute for Occupational Safety and Health
PEL	- Permissible Exposure Limit
PVC	- Polyvinyl chloride
RCRA	- Resource Conservation and Recovery Act
SARA	- Superfund Amendments and Reauthorization Act of the U.S. EPA
STEL	- Short Term Exposure Limit
TC	- Transport Canada
TDG	- Transportation of Dangerous Goods Act/Regulations
TLV	- Threshold Limit Value
TSCA	- Toxic Substances Control Act
TWA	- Time-Weighted Average
UEL	- Upper Explosive Limit

The information contained herein has been prepared by CHEMTRADE LOGISTICS Inc. and is offered only as a guide to the handling of this specific material and has been prepared in good faith by technically knowledgeable personnel. It is not intended to be all-inclusive and the manner and conditions of use and handling may involve other and additional considerations. No warranty of any kind is given or implied and CHEMTRADE LOGISTICS Inc. will not be liable for any damages, losses, injuries or consequential damages that may result from the use or reliance of any information contained herein. Chemtrade is a member of the CCPA and adheres to the codes of Responsible Care.



Prepared by Chemtrade Logistics Inc.